2,1-Benzisothiazoles. X. (1). Acylation of 2,1-Benzisothiazolin-3-one. A Remarkable Sulfur Extrusion With Ring Expansion.

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Acylation of 2,1-benzisothiazolin-3-ones (1) in pyridine gives O-acyl derivatives initially; these rapidly extrude sulfur and form substituted 3,1-benzoxazinones (4).

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We have shown previously (2) that the keto form 2,1-benzisothiazolin-3-one (1) is the preferred tautomer in the equilibrium between 1 and its enol 2, and that alkylation of the anion of 1 gives N-alkylated products.

We now find that acylation of 1 in pyridine solution affords O-acylated derivatives (3) which then extrude sulfur and rearrange to benzoxazinones (4).

The rearrangement of the O-acetyl derivatives (3); $R' = CH_3$) can be conveniently followed by nmr. On addition of acetic anhydride to a pyridine solution of 1, the methyl group signal of the O-acetyl derivative immediately appears at 2.33 δ ; within a few seconds a second peak at 2.21 δ appears and increases in intensity as the 2.33 δ peak decays. This 2.21 δ peak is due to a benzoxazinone (4). The half-lives of the O-acetyl derivatives 3a, 3b and 3c in pyridine are about 20 minutes, δ minutes, and 90 seconds, respectively, at the probe temperature of 37°.

The O-benzoyl derivatives (3, R' = Ph) rearrange in a similar fashion, but more slowly, and it is possible to isolate the O-benzoyl derivative 1a (R' = Ph). This is stable in the absence of base, but on being dissolved in pyridine and set aside overnight it deposits a mixture of sulfur and

the benzoxazinone (4a, R' = Ph). Heating of the compound to 250-300° for a few minutes in the absence of pyridine brings about the same reaction.

We suggest that the room temperature rearrangement of 3 begins by a solvent-induced ionization followed by

3 pyridine
$$R$$
 $+ R'CO^{+}N$ $+ R'CO^{+}N$

slow N-acylation of anion (5) to an N-acyl derivative 7 in which the S-N bond is so weakened that sulfur extrusion, and subsequent recyclisation to benzoxazinone (4), occurs readily.

Acylation of the 2,1-benzisothiazolin-3-ones (1) in the presence of excess aqueous base produces salts of N-acylanthranilic acids, as might be expected from the known hydrolysis of benzoxazinones (4).

EXPERIMENTAL

2,1-Benzisothiazolin-3-one (1a) and 5-Chloro-2,1-benzisothiazolin-3-one (1b).

These compounds were prepared as described previously (2). 5-Nitro-2,1-benzisothiazolin-3-one (1c).

This compound was prepared from 5-nitroisatoic anhydride by a similar procedure (3) and formed yellow leaflets (from acetic acid), m.p. 214-216°.

Anal. Calcd. for C₇H₄N₂O₃S: C, 42.85; H, 2.06; N, 14.28. Found: C, 42.88; H, 2.12; N, 14.40.

2,1-Benzisothiazol-3-yl Benzoate (3a, R' = Ph).

This compound was prepared by adding benzoyl chloride (0.3 g., 2 mmoles) to a solution of 2,1-benzisothiazolin-3-one (1a) (0.3 g., 2 mmoles) in dry pyridine (3 ml.). After 10 minutes the mixture was diluted with water (20 ml.) and the crude O-benzoate

collected and recrystallised from methanol. It formed colorless needles (0.3 g., 59%) m.p. 104° ; ir: ν (C=0) 1710 cm⁻¹; λ max 238 nm (ϵ max 26,400), 274 (15,600), 284 (17,200), 296 (15,200), 320 (11,200) and 336 (8,000). The bands at 284, 296 and 336 nm are characteristic of o-quinonoid 2,1-benzisothiazoles (2).

Anal. Calcd. for $C_{14}H_9NO_2S$: C, 65.86; H, 3.56; N, 5.49. Found: C, 65.91; H, 3.56; N, 5.44.

If the initial pyridine solution was allowed to stand overnight then sulfur was deposited together with 2-phenyl-3,1-benzoxazinone (4a; R' = Ph), m.p. and mixed m.p. 123° (lit. (4) 123°). A similar reaction occurred when isolated O-benzoate (3a, R = Ph) was redissolved in pyridine. The other 2,1-benzisothiazolin-3-ones (1b and 1c) gave the corresponding benzoxazinones 3b and 3c in similar fashion (3b, m.p. 196° [lit. (5) 196°]; 3c, m.p. 169° [lit. (6) 168°]).

Nmr Studies.

A solution of the 2,1-benzisothiazolin-3-one (1 mmole) in dry pyridine (2 ml.) containing TMS (1%) was treated with acetic anhydride (102 mg., 1 mmole). A sample of the mixture was at

once put into an nmr tube and the methyl group resonances at 2.1-2.4 δ scanned repeatedly. The nmr samples, when left overnight, deposited sulfur. In the case of the 5-nitro compound, crystalline 2-methyl-5-nitro-3,1-benzoxazinone (4c, R' = CH₃), m.p. and mixed m.p. 160° (lit. (7) $161-162^{\circ}$) was obtained.

REFERENCES AND NOTES

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